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Answers 1)  $d = 40.5 \text{ g} / 15.0 \text{ mL}$   $d = 2.7$

$\text{g/mL}$  2)  $d = 305.4 \text{ g} / 22.5 \text{ mL}$   $d =$

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13.57333 g/mL 3)  $d = g / mL$   $g = (d) (mL)$   
 $d = 2.70 \text{ g/mL}$   $d = 13.6 \text{ g/mL}$   $g = (0.789$   
 $\text{g/mL}) (200.0 \text{ mL}) = 158 \text{ g}$  4)  $(8.4 \text{ cm}) (5.5$   
 $\text{cm}) (4.6 \text{ cm}) = 212.52 \text{ cm}^3$   $d = 1896 \text{ g} /$   
 $212.52 \text{ cm}^3 = 8.9 \text{ g/cm}^3$  Significant  
figures in the answer are dictated by the  
length measurements of two sig figs. 5)  
mass of  $\text{CCl}_4 = 703.55 \text{ g} - 345.8 \text{ g} =$

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$$357.75 \text{ g} \div 225 \text{ mL} = 1.59$$

$$\text{g/mL 6) } d = 65.14 \text{ g} / 35.4 \text{ mL } d = 1.84$$

$$\text{g/mL 7) } d = \text{g} / \text{mL } \text{g} = (d) (\text{mL} \dots$$

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Density Practice Problems. 1. A platinum  
bar measures 5.0 cm long, 4.0 cm wide,



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and 1.5 cm thick. It has a mass of 700.0 grams. a. Calculate the volume of the platinum bar.  $5 \times 4 \times 1.5 = 30$  ml. b....

Density Practice Problems-KEY - Google Docs

Density = Mass  $\div$  Volume  $D = \frac{m}{V}$ . Use the density formula to solve the following

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problems. Show all work and the answer must have the correct units. Remember that volume can have different forms. A block of ice with a volume of  $3 \text{ cm}^3$  would be  $3 \text{ mL}$  of liquid after being melted. 1.

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Worksheet Remember to consider  
significant figures circle final answer; be  
sure to include units 1) A block of  
aluminum occupies a volume of 15.4 L 3)  
73 L 4) 59 L 5) 0.

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$D = \frac{m}{V}$  mass/volume. Density has units of  $\text{g/cm}^3$  or  $\text{g/ml}$ , or any mass over volume unit. In science we usually use  $\text{g/mL}$  for liquids and solids, and  $\text{g/L}$  for gases. Density is an intensive physical property. It remains the same no matter the amount of substance.

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One way to do this is to mentally compare your answer against the density of water (1 gram per cubic centimeter). Light substances would float on water, so their

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density should be less than that of water.  
Heavy materials should have density  
values greater than that of water.

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The formula for density is  $D = m \div v$ ,  
where  $m$  is the mass and  $v$  is the volume.  
When written as a fraction. the  $m$  and  $v$   
line up and give us a <3 shape! While

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many struggle with calculating density (and may not like it that much), this is a little trick to make remembering density just a little bit easier!

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Density Practice Problem Worksheet

Remember to consider significant figures  
circle final answer; be sure to include units

1) A block of aluminum occupies a  
volume of 15.0 mL and a mass of 40.5 g.  
What is its density?



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Answers from density calculations  
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Density is the amount of mass per unit volume of a given object. It can be calculated using the following equation .  $d=m/v$ . If the density of the object and either mass or volume is known, by rearranging the above equation.  $m=d*v$   
 $v=m/d$ . These three equations can be

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easily remembered by using the circle  
below.

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8 worksheets found for this concept..

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