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(c) As the  
concentration of  
oligonucleotide  
increases, another

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reaction becomes possible. A double-stranded molecule with an internal loop can form: At  $25^\circ\text{C}$ ,  $K_2 = 1.00 \times 10^{-2} \text{ M}^{-1}$  (and  $K_1 = 0.86$ ). Calculate the concentrations of all three species (single strand, SS; double strand, DS; and hairpin loop, H) at equilibrium in a

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= (-1.0 mol) x  
(8.314 J K<sup>-1</sup> THE  
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(1073 K) 35 Hence

$w = -p_{\text{ex}} \Delta V$  [2.8]

$nRT$  ( $-p_{\text{ex}}$ )  $\times nRT$

mol Even if there is

no physical piston,

the gas drives back

the atmosphere. so

the work is also

The virial

expression for

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is  $+ - 89.9$  mol

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